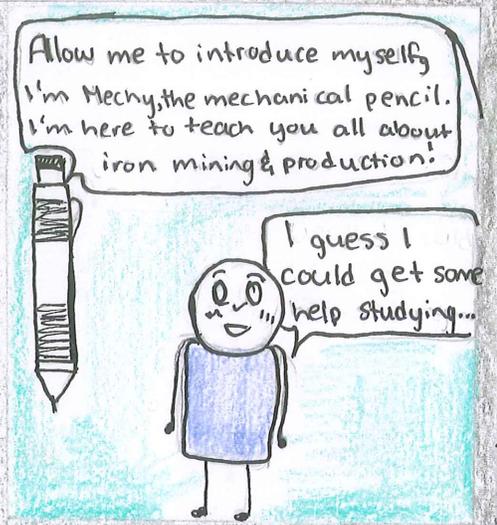
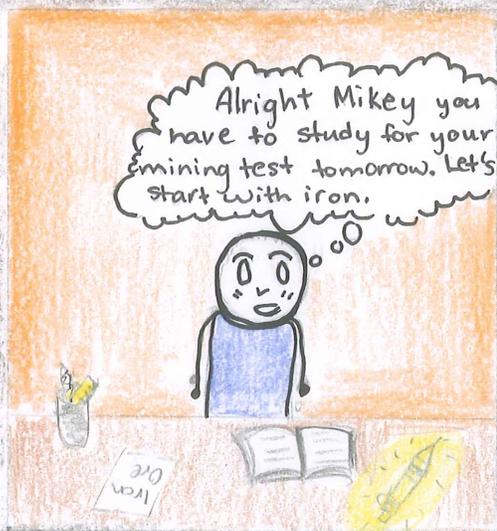


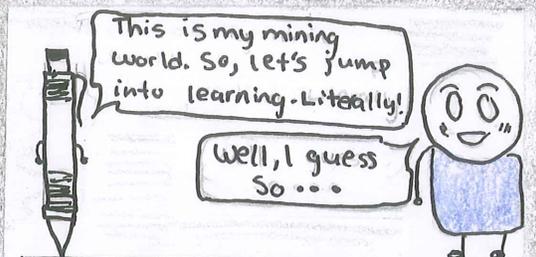
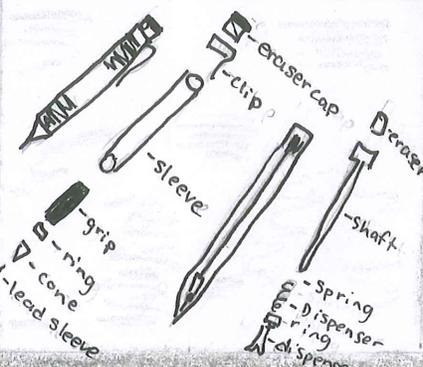


The Mining
Mechanical Pencil

By: Layla Magzoub



Before we begin we should find out what's inside Mechy.



Iron occurs in igneous, sedimentary & metamorphic rock.

Some of the most important iron bearing minerals are oxides. Magnetite, limonite, bog-iron-ore, siderite and most importantly -

Red-ish Brown Colour

- Hematite

Hematite [Fe₂O₃]

Crystalline Structure

$$\begin{array}{c} \text{O} \\ \parallel \\ \text{Fe} - \text{O} - \text{Fe} \\ \parallel \\ \text{O} \end{array}$$

Hematite



In Canada iron can be found in the Labrador Trough region. Along the border between Quebec, Newfoundland & Labrador and Nunavut.

There are four major iron mines in Canada.

1. Lac Tio	2. Mont Wright Fire Lake
3. Carol Lake	4. Bloom Lake

Company - Rio Tinto Fer et Titane	1. Quebec Lac Tio, Harve-Saint Pierre	2. Company - ArcelorMittal Mont Wright Fire Lake
Company - Iron Ore Canada	3. Newfoundland & Labrador Carol Lake	4. Company - Champion Iron Quebec Bloom Lake

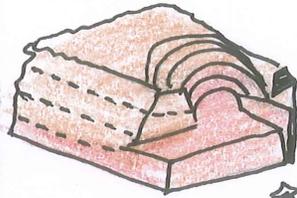
Most iron is extracted through surface mining, in open pit mines.

Now can we learn about the mining process?



About time you asked!

1. Surface Layer called the over burden is removed, creating an open pit mine.



Open Pit mine

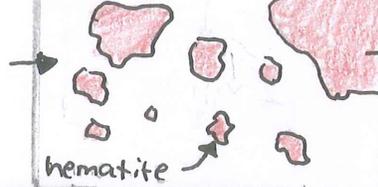


We shrank how cool!

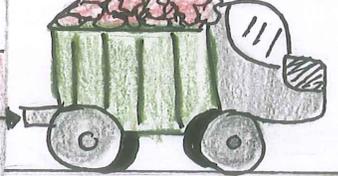


Quality of my mining world

2. Holes are drilled into the pit. Explosives are then placed to blast surrounding rock to find hematite iron ore.



3. Trucks, are loaded with the hematite and it's taken for refinery.

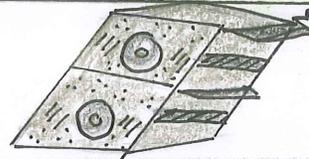
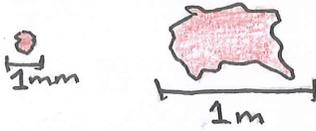
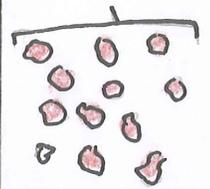


Crushing →

Mined ore contains lumps of all sizes. As big as a meter or as small as a millimeter. When we process iron ore into steel it is put through a device called a blast furnace, which can only take pieces that are 7-25 mm.

We use a process called crushing to separate iron ore lumps. Iron ore is passed through a sieve in which undersized minerals fall. If the lump is of appropriate quality it can go straight to the blast furnace. If not it must go through a process called sintering.

separated ore



Screen Sieve for stone/iron ore crushing

Now we can learn about iron ore processing.



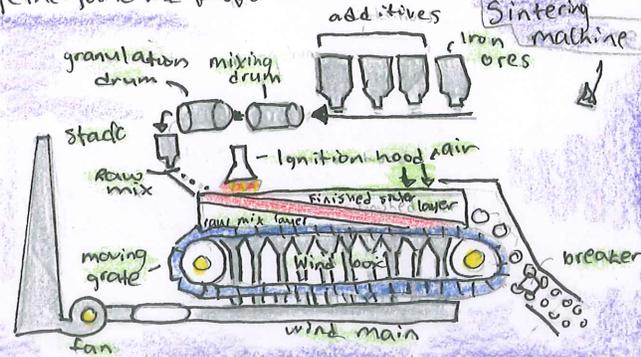
Can't wait!

Sintering

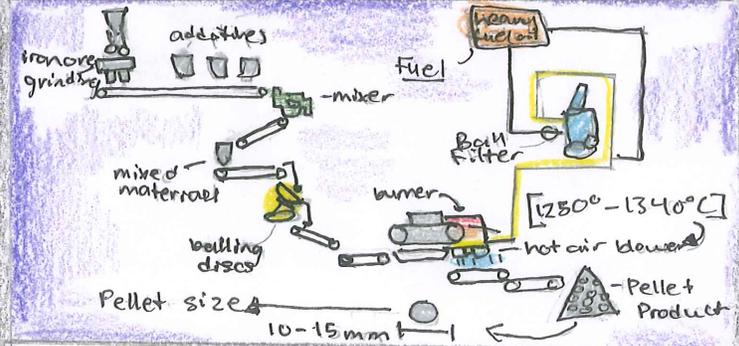
Sintering is the process of heating a layer of ore until partial melting occurs, and individual ore particles fuse together.

○ A traveling grate machine is used and the burning of a reducing agent and fuel in melting iron ore called coke is used to generate heat.

○ Before the ore is sent to a sinter machine the mixture is moistened so smaller particles stick together, and the proper amount of coke is added.



Concentrates → Upgrading & Pelletizing



Upgrading is the process of producing ore richer in iron and lower in silica. The ore is crushed and ground to release ore mineral from the valueless material where minerals are found called "gangue." Magnetic techniques are also used to separate the mineral and gangue.

Upgraded ore is in the form of a fine powder. Which is not suitable for the blast furnace, so pelletizing is used to gather the ore. Moistened concentrates are put into a rotating drum which creates soft round agglomerates. The balls are hardened by firing air.

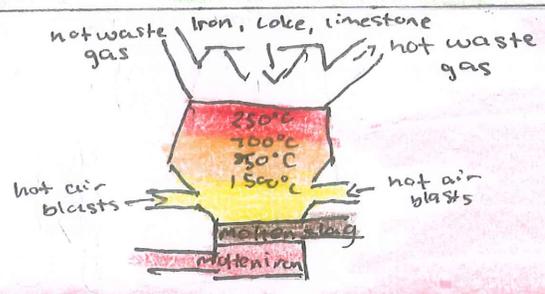
The very last step in iron processing is the b-

The blast furnace is a counter current heat and oxygen exchanger. Rising combustion gas loses most of its heat on the way up. Leaving the furnace temperature at 200°C while descending iron oxides are wholly converted into metallic iron.

-blast furnace



Blast Furnace
Iron → steel



Ah finally!! I'm going to ace my test!

We aren't done just yet...

Now that you know all this, why does it even matter?

Iron is one of the most important metals on earth. It is used in almost everything we use daily. Methyl, your barrel, springs, cone tip and lead sleeve are all made of iron.



Without iron we wouldn't have computers, or electronic devices, homes and buildings. We even need it in our bodies, our blood. We wouldn't survive without iron.

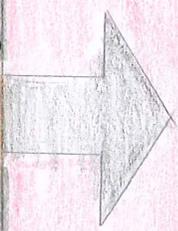
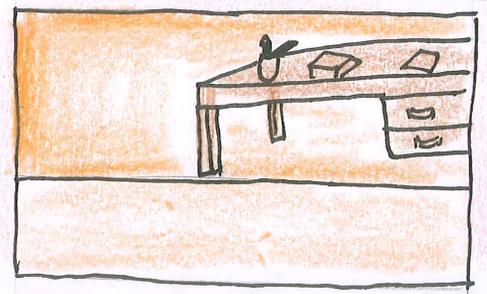
you're 100% correct none of these things would exist without iron.



well looks like it's good bye!

Thanks Methyl!

Finally back home. Now all I have to study are copper and zinc...





☺ THE END ☺

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